

Jian-Bing Liu, Hong Dai, Li-Chun Li, Wei-Feng Tao and Jian-Xin Fang\*

State Key Laboratory and Institute of Elemento-Organic Chemistry, Nankai University, Tianjin 300071, People's Republic of China

Correspondence e-mail: liu\_jianbing@sina.com

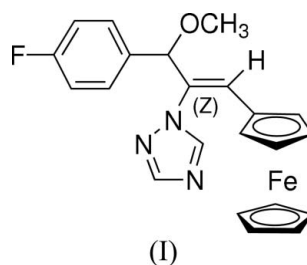
## Key indicators

Single-crystal X-ray study  
 $T = 294$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.006$  Å  
 $R$  factor = 0.042  
 $wR$  factor = 0.099  
Data-to-parameter ratio = 13.5For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.1-[(*Z*)-3-Ferrocenyl-1-(4-fluorophenyl)-1-methoxyprop-2-en-2-yl]-1*H*-1,2,4-triazole

The title compound,  $[\text{Fe}(\text{C}_5\text{H}_5)(\text{C}_{17}\text{H}_{15}\text{FN}_3\text{O})]$ , was synthesized unexpectedly during a search for potent fungicidal and plant growth regulatory agents. In the crystal structure, the molecules form centrosymmetric dimers *via* weak intermolecular  $\text{C}-\text{H}\cdots\text{F}$  interactions. The planes of the substituted cyclopentadienyl and benzene rings make dihedral angles of  $93.9$  (3) and  $15.1$  (2) $^\circ$ , respectively, with the triazole ring.

## Comment

Ferrocene is stable, non-toxic, can cross cell membranes (Dombrowski *et al.*, 1986; Yeav, 1969), and its derivatives offer exciting possibilities in insecticide and drug development (Motohashi *et al.*, 1990; Biot *et al.*, 2000). There are many studies reporting that higher activity can be achieved by introducing the ferrocenyl group into biologically active compounds (Huang & Wang, 2001; Molina *et al.*, 1999; Fang *et al.*, 2003). However, the concept of introducing ferrocenyl into triazole derivatives, which possess fungicidal, insecticidal, herbicidal and plant growth regulatory activities (Czollner *et al.*, 1990; Chu *et al.*, 1999), has not received much attention. In order to investigate novel biological triazole molecules containing the ferrocenyl group, we designed and synthesized some 1-aryl-3-ferrocenyl-2-(1*H*-1,2,4-triazole-1-yl)prop-2-en-1-ol derivatives. In this study, the crystal structure of an unexpected product, the title compound, (I), was determined.

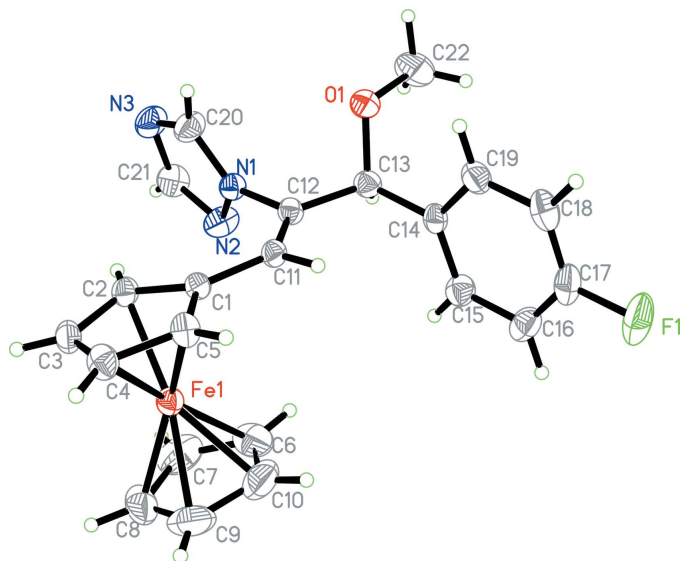


The molecular structure of (I) is shown in Fig. 1. The molecule contains four essentially planar ring systems, *viz.* two cyclopentadienyl rings [ $p1$  (C1–C5) and  $p2$  (C6–C10)], a triazole ring ( $p3$ ) and a substituted benzene ring ( $p4$ ). The dihedral angles between  $p3$  and  $p1$ , and between  $p3$  and  $p4$ , are  $93.9$  (3) and  $15.1$  (2) $^\circ$ , respectively. The  $\text{Fe}-\text{C}$  distances in the ferrocenyl ring are normal and the distance of the cyclopentadienyl ring centroids  $\text{Cg}1$  (C1–C5) and  $\text{Cg}2$  (C6–C10) from atom  $\text{Fe}1$  are not significantly different [1.638 (2) and 1.650 (3) Å, respectively]. The cyclopentadienyl rings are slightly staggered, as evidenced by the  $\text{C}1\cdots\text{Cg}1\cdots\text{Cg}2\cdots\text{C}6$  and  $\text{C}3\cdots\text{Cg}1\cdots\text{Cg}2\cdots\text{C}8$  pseudo-torsion angles of  $-12.0$  (2) and  $-14.0$  (3) $^\circ$ , respectively.

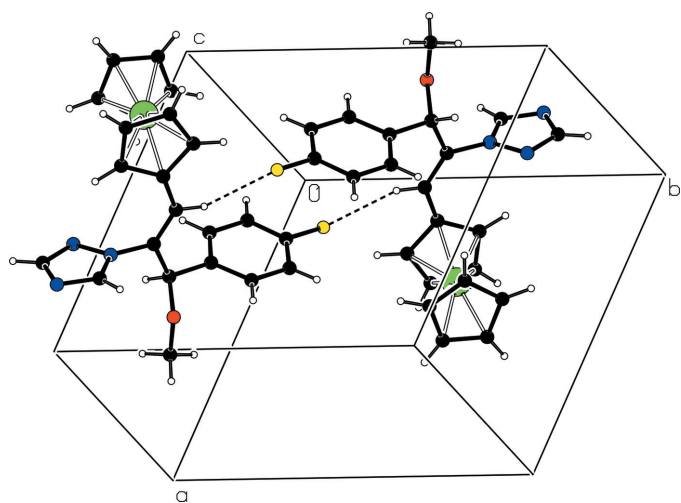
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**Figure 1**  
A view of (I), with displacement ellipsoids drawn at the 30% probability level.



**Figure 2**  
A view (Spek, 2003) of a centrosymmetric dimer of (I). Dashed lines indicate weak C—H...F interactions.

In the crystal structure of (I), weak intermolecular C—H...F interactions [ $H \cdots F^i = 2.511 \text{ \AA}$ ,  $C \cdots F^i = 3.386(3) \text{ \AA}$  and  $C11-H11 \cdots F^i = 156.8(2)^\circ$ ; symmetry code: (i)  $1 - x, 1 - y, 2 - z$ ] link the molecules into centrosymmetric dimers (Fig. 2).

## Experimental

1-(4-Fluorophenyl)-3-ferrocenyl-2-(1*H*-1,2,4-triazol-1-yl)prop-2-en-1-one (4.2 g, 10 mmol) was dissolved in methanol (15 ml) and water (20 ml). Sodium borohydride (0.076 g, 20 mmol) was then added in six batches below 283 K. The mixture was stirred for 24 h at room temperature, then adjusted to pH 6 using 10% (w/w) sulfuric acid. The solution was extracted with diethyl ether ( $3 \times 20 \text{ ml}$ ), and the combined organic layer was washed with water ( $3 \times 20 \text{ ml}$ ) and then dried over anhydrous magnesium sulfate. After removal of the solvent, the residue was recrystallized from petroleum ether–ethyl acetate (4:1 v/v) to give yellow crystals of (I) (yield 56%).

## Crystal data

$[\text{Fe}(\text{C}_5\text{H}_5)(\text{C}_{17}\text{H}_{15}\text{FN}_3\text{O})]$   
 $M_r = 417.26$   
 Triclinic,  $P\bar{1}$   
 $a = 10.116(2) \text{ \AA}$   
 $b = 11.045(3) \text{ \AA}$   
 $c = 11.239(3) \text{ \AA}$   
 $\alpha = 100.554(4)^\circ$   
 $\beta = 110.267(4)^\circ$   
 $\gamma = 115.736(4)^\circ$   
 $V = 975.0(4) \text{ \AA}^3$

$Z = 2$   
 $D_x = 1.421 \text{ Mg m}^{-3}$   
 Mo  $K\alpha$  radiation  
 Cell parameters from 1579 reflections  
 $\theta = 2.2\text{--}22.8^\circ$   
 $\mu = 0.80 \text{ mm}^{-1}$   
 $T = 294(2) \text{ K}$   
 Block, yellow  
 $0.24 \times 0.22 \times 0.18 \text{ mm}$

## Data collection

Bruker SMART CCD area-detector diffractometer  
 $\varphi$  and  $\omega$  scans  
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)  
 $T_{\min} = 0.821$ ,  $T_{\max} = 0.866$   
 4998 measured reflections

3417 independent reflections  
 2445 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.025$   
 $\theta_{\max} = 25.0^\circ$   
 $h = -12 \rightarrow 10$   
 $k = -7 \rightarrow 13$   
 $l = -12 \rightarrow 13$

## Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.042$   
 $wR(F^2) = 0.099$   
 $S = 1.02$   
 3417 reflections  
 254 parameters

H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0447P)^2]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.004$   
 $\Delta\rho_{\max} = 0.25 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\min} = -0.24 \text{ e \AA}^{-3}$

**Table 1**

Selected geometric parameters ( $\text{\AA}$ ,  $^\circ$ ).

N1—C20	1.319 (4)	O1—C22	1.428 (4)
N1—N2	1.360 (3)	C11—C12	1.321 (4)
N1—C12	1.440 (3)	C12—C13	1.510 (4)
N2—C21	1.317 (4)	C13—C14	1.513 (4)
N3—C20	1.311 (4)	C14—C19	1.380 (4)
N3—C21	1.338 (4)	C14—C15	1.383 (4)
O1—C13	1.417 (3)		
C20—N1—N2	108.8 (2)	C11—C12—N1	121.1 (2)
C20—N1—C12	129.2 (3)	C11—C12—C13	126.8 (3)
N2—N1—C12	121.7 (2)	N1—C12—C13	112.2 (2)
C21—N2—N1	101.5 (3)	O1—C13—C12	105.2 (2)
C20—N3—C21	101.3 (3)	O1—C13—C14	112.6 (2)
C13—O1—C22	113.6 (2)	C12—C13—C14	113.9 (2)
C2—C1—C11	130.9 (3)	N3—C20—N1	112.2 (3)
C5—C1—C11	122.5 (3)	N2—C21—N3	116.2 (3)
C20—N1—N2—C21	−0.2 (3)	C12—N1—N2—C21	−175.1 (3)

All H atoms were placed in calculated positions, with C—H = 0.93  $\text{\AA}$ , or 0.96  $\text{\AA}$  for methyl H, and included in the refinement using a riding-model approximation, with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ , or  $1.5U_{\text{eq}}(\text{C})$  for methyl H.

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1999); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1999); software used to prepare material for publication: SHELXTL.

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## References

- Biot, C., Delhaes, L., Lucien, A. M., Mortuaire, M., Camus, D., Divd, S. & Brocard, S. S. (2000). *Eur. J. Med. Chem.* **35**, 707–714.
- Bruker (1998). *SMART*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Bruker (1999). *SAINT* and *SHELXTL*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Chu, C. H., Sun, X. W., Sun, L., Zhang, Z. Y., Li, Z. C. & Liao, R. A. (1999). *J. Chin. Chem. Soc.* **46**, 229–235.
- Czollner, L., Sxilagli, G. & Janaky, J. (1990). *Arch. Pharm. (Weinheim)*, **323**, 225–229.
- Dombrowsk, K. E., Baldwin, W. & Sheats, J. E. (1986). *J. Organomet. Chem.* **302**, 281–306.
- Fang, J. X., Jin, Z., Liu, Z. & Liu, W. (2003). *J. Organomet. Chem.* **674**, 1–9.
- Huang, R. Q. & Wang, Q. M. (2001). *J. Organomet. Chem.* **94**, 637–639.
- Molina, P., Tarraga, A., Lopez, J. L. & Martinez, J. C. (1999). *J. Organomet. Chem.* **584**, 147–158.
- Motohashi, N., Meyer, R., Gollapudi, S. R. & Bhattiprolu, K. R. (1990). *J. Organomet. Chem.* **398**, 205–217.
- Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.
- Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
- Yeav, R. A. (1969). *Toxicol. Appl. Pharmacol.* **15**, 666–673.